

Preparation of New Chiral Peralkyldiphosphines as Efficient Ligands for Catalytic Asymmetric Hydrogenation of α -Dicarbonyl Compounds

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New chiral peralkyldiphosphines (**5**) containing a pyrrolidine ring have been prepared by a new method, which is widely applicable to transformation of known chiral diphosphines into the corresponding cyclohexyl analogues; asymmetric hydrogenation of α -dicarbonyl compounds catalysed by rhodium complexes of (**5**) proceeds smoothly under mild conditions (1 atm H₂; 35 °C) to give moderate optical yields.

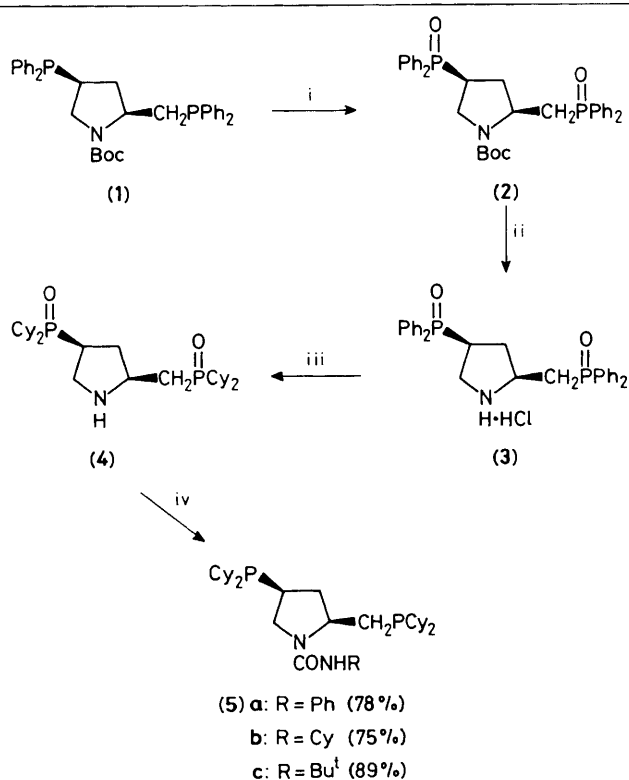
There are only a limited number of reports on the catalytic asymmetric hydrogenation of ketones compared to that of olefins.¹ One reason seems to be that few efficient catalysts are available; ketones are not reduced with classical Wilkinson catalysts. Recently we have found that rhodium(I) peralkyldiphosphine complexes show high catalytic activity for hydrogenation of various kinds of ketones.² As the first

example of optically active peralkyldiphosphines which may be used in the catalytic asymmetric hydrogenation of prochiral ketones we have prepared the ligand Rdiop [2,3-*O*-isopropylidene-2,3-dihydroxy-1,4-bis(dialkylphosphino)-butane] and showed that its rhodium(I) complexes are effective catalysts for asymmetric hydrogenation of ketones.³ We now report the preparation of the new chiral peralkyldi-

Table 1. Physical properties of the chiral peralkyldiphosphines (**5**).

Compound	M.p. (<i>t</i> /°C)	I.r. (ν/cm^{-1})			$^{31}\text{P}\{^1\text{H}\}$ N.m.r. ^a $\delta(\text{CHCl}_3)$	$[\alpha]_{\text{D}}^{21}(\text{C}_6\text{H}_6)$
		$\nu(\text{NH})$	$\nu(\text{CO})$	Amide		
(5a)	212—212.5	3300	1645	1535	+6.9(s), -11.1(s)	-30.4° (c 0.68)
(5b)	192.5—194	3260	1615	1540	+5.6(s), -11.6(s)	-13.9° (c 0.96)
(5c)	166—168	3290	1625	1530	+6.4(s), -12.3(s)	-13.3° (c 1.62)

^a In p.p.m. from external H_3PO_4 ; downfield is positive.

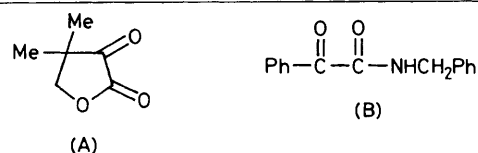


Cy = cyclohexyl; Boc = *t*-butoxycarbonyl

Scheme 1. Reagents: i, 10% H_2O_2 , acetone, 85% yield; ii, 25% HCl , MeOH , 86% yield; iii, H_2 , $\text{Rh-Al}_2\text{O}_3$, then NaOH , 73% yield; iv, HSiCl_3 , NEt_3 , then RNCO .

phosphines (**5**) containing a pyrrolidine ring by a new method (Scheme 1),[†] which may be widely applicable to the transformation of known chiral aryldiphosphines into the corresponding cyclohexyl analogues. Optically active diphosphines containing a pyrrolidine ring are known to be excellent ligands for asymmetric catalysis and show a much higher degree of asymmetric induction in the hydrogenation of dehydroamino-acid derivatives than diop.^{4,5} The diphosphines (**5**) prepared here are thus expected to be more efficient ligands in asymmetric hydrogenation of ketones than *R*diop.

Conventional methods [reaction between the alkali metal dialkylphosphide and the corresponding chiral bis-toluene-*p*-sulphonate], could not be used for the preparation of (**5**) (or *R*diop^{3a}). Attempts to convert the pyrrolidine derivative (**1**) into the cyclohexyl analogue by direct hydrogenation of the diphenylphosphino groups failed. We hoped to prepare the cyclohexyl compounds (**5**)⁵ via hydrogenation of diphenylphosphinoethyl groups to give dicyclohexylphosphinoethyl groups.⁶ Oxidation of (**1**) with 10% hydrogen peroxide in ice-cooled acetone proceeded smoothly during 2 h to give the *P,P'*-

**Table 2.** Asymmetric hydrogenation of α -dicarbonyl compounds catalysed by Rh -(**5**) complexes.^a

Substrate	Ligand	$t_{1/2}/\text{min}^b$	% Enantiomeric excess ^c (Config.)
(A)	(5a)	26	41(<i>S</i>)
	(5b)	23	50(<i>S</i>)
	(5c)	16	62(<i>S</i>)
	(5c) ^d	20	66(<i>S</i>)
(B)	(5a)	19	35(<i>R</i>)
	(5b)	20	41(<i>R</i>)
	(5c)	25	47(<i>R</i>)

^a All reactions were carried out with [substrate] = 0.5 M, $[\text{Rh}] = 2.5 \text{ mM}$ (0.5 mol %) under atmospheric pressure of hydrogen at 35 °C in tetrahydrofuran unless otherwise stated. The catalyst precursors were prepared *in situ* from $[\text{Rh}(\text{C}_8\text{H}_{14})_2\text{Cl}]_2$ and 2.2 equiv. of (**5**). ^b Times required for 50% conversion. ^c Optical yields were determined on the basis of the maximum rotations of the pure enantiomers: *D*-pantoyl lactone, $[\alpha]_{\text{D}}^{25} - 50.7^\circ$ (c 2.05, H_2O);⁸ (*S*)-*N*-benzylmandelamide, $[\alpha]_{\text{D}}^{25} + 79.9^\circ$ (c 1.09, CHCl_3).^{3c,d} Benzene was used as a solvent.

dioxide (**2**), m.p. 227—230 °C, $[\alpha]_{\text{D}}^{21} - 12.6^\circ$ (c 1.18, benzene). Removal of the Boc group by treating (**2**) with methanolic hydrogen chloride at 0 °C gave the bis-diphenylphosphinoethyl compound (**3**), m.p. 268—270 °C, $[\alpha]_{\text{D}}^{21} + 5.9^\circ$ (c 2.69, MeOH), which was hydrogenated to provide the cyclohexyl analogue. The hydrogenation was somewhat sluggish, and $\text{Rh-Al}_2\text{O}_3$ was the most satisfactory catalyst. The bis-diphenylphosphinoethyl compound (**3**) was hydrogenated in methanol with hydrogen (130 atm) at 150 °C for 2 days over 5% $\text{Rh-Al}_2\text{O}_3$. After neutralisation and recrystallisation from $\text{CHCl}_3\text{-Et}_2\text{O}$ the bis-dicyclohexylphosphinoethyl analogue (**4**) was isolated as colourless crystals containing 2 moles of solvating chloroform, m.p. 215—215.5 °C, $[\alpha]_{\text{D}}^{21} - 11.4^\circ$ (c 2.30, benzene). Reduction of the phosphine oxide was achieved by refluxing (**4**) with $\text{HSiCl}_3\text{-NEt}_3$ in acetonitrile under nitrogen for 2 h and followed by treatment with 25% NaOH .⁷ The resulting oily product, *N*-unsubstituted diphosphine, without purification, was transformed into the crystalline product (**5**) by reaction with the appropriate isocyanate in CH_2Cl_2 at room temperature for 2 h. Compounds (**5a-c**) were isolated in good yields as colourless crystals after recrystallisation from ethanol and their physical properties are listed in Table 1. Two singlet $^{31}\text{P}\{^1\text{H}\}$ n.m.r. signals for each diphosphine indicate that the diphosphines (**5**) are optically pure and that racemisation did not occur during their preparation.

[†] All new compounds, (**2**)—(**5**), gave satisfactory spectroscopic data and elemental analyses.

As expected, the rhodium(I) complexes of (5) showed high catalytic activity for hydrogenation of ketones. As shown in Table 2 α -dicarbonyl compounds were hydrogenated smoothly under atmospheric pressure of hydrogen at 35 °C to give the corresponding optically active α -hydroxy carbonyl compounds in moderate optical yields.

The present findings may provide a general method of conversion of known chiral diphosphines into the corresponding cyclohexyl analogues leading to more efficient chiral ligands for ketone hydrogenation.

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